

## High-frequency EPR investigation of X-ray storage SrBPO<sub>5</sub>:Eu phosphor

T. Nakamura<sup>a,\*</sup>, T. Takeyama<sup>a</sup>, N. Takahashi<sup>a</sup>, R. Jagannathan<sup>b</sup>,  
A. Karthikeyani<sup>b</sup>, G.M. Smith<sup>c</sup>, P.C. Riedi<sup>c</sup>

<sup>a</sup>Department of Materials Science and Technology, Faculty of Engineering, Shizuoka University, 3-5-1 Johoku,  
Hamamatsu 432-8561, Japan

<sup>b</sup>Luminescence Group, Council of Science and Industrial Research, Central Electrochemical Research Institute,  
Karaikudi 630006, TN, India

<sup>c</sup>School of Physics and Astronomy, University of St. Andrews, North Haugh, St. Andrews Fife KY16 9SS, UK

### Abstract

High-frequency EPR (HFEP) spectra of the SrBPO<sub>5</sub>:Eu with stillwellite structure, which was prepared using EuCl<sub>3</sub>·6H<sub>2</sub>O as a dopant, have been examined in order to clarify the Eu sites in it. They showed an intense doublet at 3.3 T corresponding to  $g = 2.0$  as is expected for the <sup>8</sup>S state ions. Also, weak satellites were observed on both sides of the central doublet signal symmetrically. This means that microwave energy at 90 GHz is sufficiently larger than the zero-field splitting. Taking into account the crystal structure of stillwellite and the earlier PL investigation, a single Eu<sup>2+</sup> site was assumed for the HFEP simulation. As a result, it was found that HFEP spectrum simulated using  $g = 1.98161$ ,  $|D| = 0.130$  T and  $\lambda = |E/D| = 0.06$  reasonably agrees well with the experimental one, supporting that the Eu<sup>2+</sup> occupies the Sr site. The large  $|D|$  value small  $\lambda$  value obtained implies that the crystal field surrounding Eu<sup>2+</sup> has a strong axial field with weak rhombic distortion.

© 2003 Elsevier Science B.V. All rights reserved.

**Keywords:** HFEP; Europium(II); X-ray storage phosphor; Stillwellite

### 1. Introduction

X-ray storage phosphors are practically useful for digital radiography in the medical diagnosis field. Recently, two of the authors have examined Eu<sup>2+</sup>-doped phosphor SrBPO<sub>5</sub>:Eu with an emission at 390 nm by means of X-ray crystallography, X-band EPR spectroscopy and photoluminescence spectroscopy [1]. According to them, at low

loadings of Eu<sup>2+</sup> the X-band EPR spectra showed 12 lines centred around  $g = 2.0$ , which suggests that Eu<sup>2+</sup> occupies a high-symmetry site. At higher loadings, however, very complicated EPR resonance, which is not simply assigned to Eu<sup>2+</sup> in a single site, was observed, implying that some Eu<sup>2+</sup> occupy low-symmetry site(s).

Previously we have succeeded in identifying low-symmetry Eu<sup>2+</sup> which is doped in alkaline earth aluminate phosphors by means of high-frequency EPR (HFEP) spectroscopy [2–5], and confirmed that the HFEP has advantage over X-band EPR

\*Corresponding author. Tel./fax: +81-53-478-1156.

E-mail address: [tctnaka@ipc.shizuoka.ac.jp](mailto:tctnaka@ipc.shizuoka.ac.jp) (T. Nakamura).

for the investigation of metal ions with a large zero-field splitting. In this paper, therefore, we report the results of the HFEPR investigation of the SrBPO<sub>5</sub>:Eu phosphor.

## 2. Experimental

Eu<sup>2+</sup>-doped samples were prepared by conventional solid-state reaction with SrCO<sub>3</sub>, H<sub>3</sub>BO<sub>3</sub>, (NH<sub>4</sub>)HPO<sub>4</sub> and EuCl<sub>3</sub>·6H<sub>2</sub>O (or Eu<sub>2</sub>O<sub>3</sub>) under mild reducing atmosphere. Eu<sup>2+</sup> concentration was adjusted to 1 mol%, because in the previous study the optimum europium(II) concentration for photoluminescence and photo-stimulated luminescence is 1 mol% [1]. The preparation procedure was described elsewhere [1]. HFEPR measurements at 90 GHz were carried out at room temperature using a novel quasi-optical induction mode spectrometer that was designed and built at the University of St. Andrews, UK.

## 3. Results and discussion

Fig. 1 shows the X-band EPR spectrum of the SrBPO<sub>5</sub>:Eu prepared with a EuCl<sub>3</sub> of 1.0 mol%. The same spectral pattern was obtained when

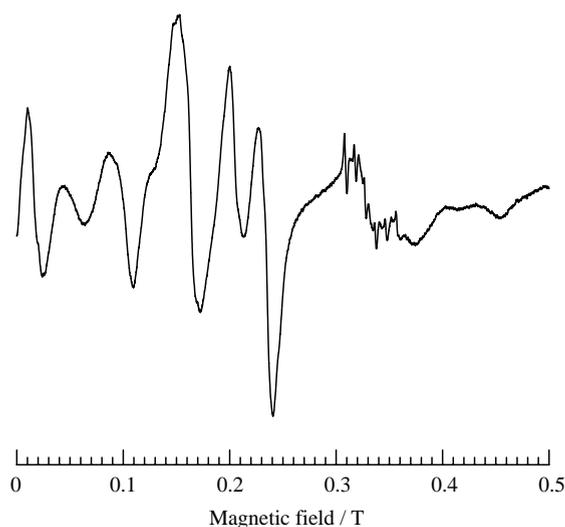


Fig. 1. X-band EPR spectrum of the SrBPO<sub>5</sub>:Eu with stillwellite structure recorded at room temperature.

Eu<sub>2</sub>O<sub>3</sub> was used as a dopant, implying that the chloride ion rarely affects the environment surrounding Eu<sup>2+</sup>. Also, it was confirmed that the EPR spectrum obtained in this study is fundamentally the same as those reported by one of the authors earlier, although the resonance attributed to Eu<sup>2+</sup> in a high-symmetry site is not obvious because of the intense hyperfine lines due to Mn<sup>2+</sup> impurity.

It is seen that several broad signals appear over the wide range of magnetic field. Amongst them the signals appearing in the magnetic field in the range of 0–0.25 T are stronger than those observed at about 0.33 T although the *g*-value of <sup>8</sup>S state ions without orbital angular momentum should be close to 2.0. Such X-band EPR spectrum is a typical pattern for the Eu<sup>2+</sup> with a large zero-field splitting, which results in the mixing of the wave functions. In other words, Eu<sup>2+</sup> occupies low-symmetry site(s) [2–5].

Fig. 2(a) shows the 90 GHz HFEPR spectrum of the SrBPO<sub>5</sub>:Eu prepared using EuCl<sub>3</sub>·6H<sub>2</sub>O as a

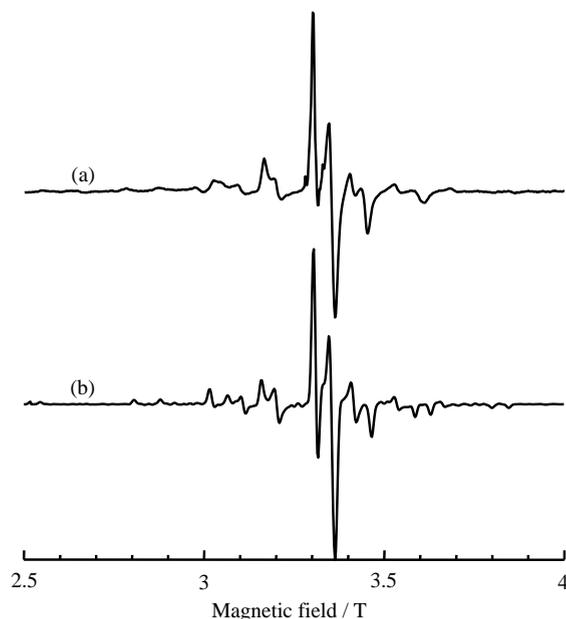


Fig. 2. Room-temperature 90 GHz EPR spectra of the SrBPO<sub>5</sub>:Eu with stillwellite structure: (a) experimental spectrum and (b) simulated spectrum with  $g = 1.9816$ ,  $|D| = 0.130$  T and  $\lambda = |E/D| = 0.06$ . Lorentzian line shape with a peak-to-peak width of  $\Delta H = 0.0075$  T was assumed.

dopant. It is immediately noticed that the obtained HFEP spectrum is different from that measured at X-band. An intense doublet signal appears at 3.3 T corresponding to  $g = 2$  as is expected for the  $^8S$  state ions. Also, weak satellites are observed on both sides of the doublet signal symmetrically. When the HFEP spectrum is measured under the condition satisfying Zeeman energy  $\gg$  zero-field splitting energy, it consists of seven fine structure lines for  $\Delta M_S = \pm 1$ . This means that microwave energy at 90 GHz is large enough compared with the zero-field splitting.

Based on the crystal structure of stillwellite [6] and the results of the photoluminescence investigation reported earlier [1] a single  $\text{Eu}^{2+}$  site is assumed for the HFEP simulation. EPR parameters were roughly estimated as follows:  $g$ -value was calculated using  $g = hv/\beta H$  with the magnetic field where the central intense signal crosses zero in intensity. From the satellites appearing at 3.19 and 3.45 T next to the intense doublet the  $|D|$  value is 0.15 T. For the  $|E/D|$  ( $= \lambda$ ) an initial value of 0.1 was used. This is because three absorption-like satellites at  $\lambda = \frac{1}{3}$  should be expected on both sides of the central signal, while three absorption-like and three derivative-like satellites at  $\lambda = 0$ . Observed HFEP spectrum implies that the  $\lambda$  value is in between.

Simulated HFEP spectrum, which was carried out using the equation derived using the second perturbation theory [7,8], is shown in Fig. 2(b). The parameters used were  $g = 1.9816$ ,  $|D| = 0.130$  T and  $\lambda = |E/D| = 0.06$ . It is obvious that the simulated HFEP spectrum reasonably agrees well with the experimental one, supporting that the  $\text{Eu}^{2+}$  occupies the strontium site since there is only a single site in the unit cell of stillwellite. The obtained  $|D|$  value is very large, while the  $\lambda$  value is small, implying that the crystal field surrounding  $\text{Eu}^{2+}$  has a strong axial field with weak rhombic distortion.

From the photoluminescence studies [1], we observe that the stillwellite-type  $\text{SrBPO}_5$  system, for low levels of  $\text{Eu}^{2+}$  doping ( $\sim 0.01\%$ ) shows a weak emission band around 350 nm while for moderate level of  $\text{Eu}^{2+}$  doping (1–5%) an intense narrow band around 390 nm is observed. These two emission bands might suggest two kinds of

cationic site(s) occupied by  $\text{Eu}^{2+}$  in this system. The absence of 350 nm emission band for higher levels of  $\text{Eu}^{2+}$  doping may be explained in terms of intersite energy transfer in favour of 390 nm emission band having profound dependence on  $\text{Eu}^{2+}$  concentration. For all practical purposes, the intense 390 nm emission band due to  $\text{Eu}^{2+}$  centre in  $\text{SrBPO}_5$  can be considered to represent  $\text{Eu}^{2+}$  emission in  $\text{SrBPO}_5$ . The band emission from  $\text{Eu}^{2+}$  centre(s) can be assigned to  $4f^65d \rightarrow ^8S_{7/2}$  (an allowed f–d type) transition. The occurrence of band emission due to parity allowed ‘f–d’ transition that is possible irrespective of whether the cationic site(s) occupied by  $\text{Eu}^{2+}$  has an inversion centre or not and hence symmetry independent.

The narrow band emission observed from  $\text{Eu}^{2+}$  centre(s) clearly point out that the lowest excited state is  $4f^65d$  state. The 5d orbital of  $\text{Eu}^{2+}$  can undergo considerable crystal field splitting determined by the symmetry and the chemical coordination of the cationic site(s) it occupies.

In conclusion, it is found that for the  $\text{SrBPO}_5:\text{Eu}$  the  $\text{Eu}^{2+}$  occupies the strontium site which has strong axial crystal field with weak rhombic distortion. This explains the PL behaviour as follows: A simple comparison of the excitation spectrum of  $\text{SrBPO}_5:\text{Eu}^{2+}$  in relation to cubic  $\text{KCl}:\text{Eu}^{2+}$  readily reveals that the excitation spectrum due to the splitting of the 5d orbital in  $\text{SrBPO}_5$  is more complex and broader by approximately 20%. This suggests that the axial field parameter  $|D| = B_2^0$  is much larger in  $\text{SrBPO}_5$ . Further, the complex excitation spectrum of  $\text{SrBPO}_5:\text{Eu}^{2+}$  implies that this cationic site can have weak rhombic field too (low value for  $E$ ). These results are possible for  $\text{Eu}^{2+}$  occupying a 9 or 10 coordinated sites due to the fact that  $\text{Sr}^{2+}$  has no symmetry elements [9,10]. These results are in good agreement with the conclusions deduced from the HFEP studies on  $\text{SrBPO}_5:\text{Eu}^{2+}$ .

## Acknowledgements

This work was supported by the Japan Society for the Promotion of Science through a Grant-in-Aid for the Scientific Research (A) No. 12355030.

**References**

- [1] A. Karthikeyani, R. Jagannathan, *J. Lumin.* 86 (2000) 79.
- [2] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *J. Chem. Soc. Faraday Trans.* 94 (1998) 3009.
- [3] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *Phys. Chem. Chem. Phys.* 1 (1999) 4011.
- [4] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *J. Mater. Chem.* 10 (2000) 2566.
- [5] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, M. Ohta, C.C. Rowlands, G.M. Smith, P.C. Riedi, *Phys. Chem. Chem. Phys.* 3 (2001) 1721.
- [6] A.A. Voronkov, Yu.A. Pyatenko, *Sov. Phys. Cryst.* 12 (1967) 214.
- [7] J.R. Pilbrow, *Transition Ion Electron Paramagnetic Resonance*, Clarendon Press, Oxford, 1990, p. 614.
- [8] T. Takeyama, T. Nakamura, T. Takahashi, V. Beltran-Lopez, C.C. Rowlands, in: A. Kawamori, J. Yamauchi, H. Ohta (Eds.), *EPR in the 21st Century*, Elsevier Science B.V., Netherlands, 2001, 63pp.
- [9] Y. Ono, K. Takayanama, T. Kajitani, *J. Phys. Soc. Jpn.* 65 (1996) 3224.
- [10] A.A. Voronkov, Yu.A. Pyatenko, *Kristallografiya* 12 (1967) 135.