An Approach to Corrosion Auditing in Indian Sugar **Factories**

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in a sugar factory there are many components that can undergo corrosion. The sugar industry is the second-largest agro-processing industry in India. A cost of corrosion survey was conducted in a few sugar factories by using the present-value method. It concludes that the direct annual cost of corrosion in the sugar industry as a whole is ~U.S. \$14 million. It also shows that nearly U.S. \$2.2 million per year could be saved by using type 409M stainless steel (UNS \$40900) in place of ordinary mild steel.

> orrosion always has been recognized as one of the most serious problems affecting every core industrial sector, and the annual loss from corrosion is great. Studies undertaken in the U.S., the U.K., Japan, Australia, Kuwait, Germany, Finland, Sweden, India, and China have shown an annual cost ranging from ~I to 6% of the Gross National Product (GNP) of any given country. For India, the cost was ~2% of the GNP for the year 1984 to 1985.1 A recent study2 estimated that the to

tal cost of corrosion in the U.S billion (3.1% of the GNP).

Different approaches exist the cost of corrosion. The auth fer the present-value-or dis cash flow-method. This to converts all expenditures as with corrosion into equivalent costs, using the sum of these costs to arrive at the annual co rosion. This article discusses th this approach to determine the corrosion in the Indian sugar in

The sugar industry is the largest agricultural processing in India and is of considerable tance to rural economies. The has 506 factories, with a total of 11 million tons (10 billion k

According to the Indian Sug Association (New Delhi, Ind. sugar industry's total investmen billion and its total annual si \$4.2 billion.

Sugar production consists of of processes that can be row vided into four major manufa stages:

- Stage 1—Cane to raw juid
- Stage 2—Raw juice to d
- Stage 3—Clarified juice to
- Stage 4—Syrup to sugar.

This article highlights the cor problems faced by the sugar in and the costs it must bear com them. Table I lists the corrosion lems encountered at each stag sugar factory, there is a large of components-all of which dergo erosion-corrosion (abrasiv combined with corrosion). Bo the material of construction is: mild steel, general corrosion at sion-corrosion are common a nance problems. In some instance alternative, cost-effective, and me rable construction material ma an economical solution.

Corrosion Auditing

Each year, the sugar industr mally spends a considerable st

DESCRIPTION OF PLANT COMPONENTS, FUNCTION, AND CORROSION PROBLEMS OF A TYPICAL SUGAR FACTORY

lame of the System	Parts of the System	Material of Construction	Function	Typical Corrosion
ar unloader	Electrically operated trolley, column structure, wheels, and rails	Mild and cast steel	Transport of sugar cane from lorries and trucks to feeder table.	General and wear corrosion
nder table	Structurals, chains, and drives	Mild steel	Sugar cane is dumped here for further processing.	General corrosion
e carrier	Structurals, chains, slats, and drives	Mild steel	Transport of sugar canes to the milling section	General and wear corrosion
e kicker	Drum with steel blades	Mild steel	Sugar cane is shredded into fine pieces for easy squeezing.	Acid attack, general and wear corrosion
eleveller	Knife, hub, hood (top cover), and drives	Mild and cast steel	Sugar cane is shredded into fine pieces for easy squeezing.	Acid attack, general and wear corrosion
fibrizer	Hammer, hub, anvil plate, hood, and drives	Cast and mild steel	Sugar cane is shredded into fine pieces for easy squeezing.	Acid attack, general and wear corrosion
emilling on	Cheek, trash beams, trash plate, scrubbers, roller shaft roller shell, Messachesert knife, roller bearings, under- feeder roller, intercarrier, juice pipes and tanks	Cast iron, mild steel, forged iron, and SS.	Cane is crushed here to extract the raw juice.	General, wear, acid attack, and pitting corrosion
filter screen	Filter screen	Mild steel	Removal of bagasse and fine particles from the raw juice.	Wear corrosion
collecting	Tank	Mild steel	Used to collect raw juice.	Acid attack
weighing tank	Tank	Mild steel	Raw juice is weighed.	Acid attack
receiving	Tank	Mild steel	Weighed juice is received here for further processing.	General corrosion and acid attack
carrying	Pipes	Mild steel	To carry the raw juice from juice receiving tank to juice clarifier.	Wear corrosion and acid attack
ration unit	Lime slaker, lime classifier, and lime stirrer.	Mild steel	Preparation of milk of lime and pumping into juice sulfitation station.	Alkaline attack
cation unit	Melter, melt receiving chamber, and vapor pressure chamber (furnace)	Mild steel and mild steel with steam jacket.	Preparation of sulfur dioxide (SO ₂) gas for sulfitation.	SO_2 gas attack
once	Shell, tubes, and valves	Cast iron, mild steel, and SS	For settlement of impurities.	Scale formation
sulfiter	Cylindrical wessel with stirrer	Mild steel	Simultaneous mixing of raw juice, lime, and SO ₂ .	SO ₂ gas attack

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Materials Selection & Design

TABLE 1 (CONTINUED)

Name of the System	Parts of the System	Material of Construction	Function	Typical Corrosion
Secondary juice heater	Shell, tubes, and valves	Cast iron, mild steel, and SS	To reduce the water content in the juice.	Deposition of sludge
Juice clarifier		Mild steel	For separation of clear juice from precipitated impurities (mud).	Acid attack and wear corrosion
Mud collecting tank (or) mud mixer	Tank	Mild steel	Precipitated impurities at the clarifier are collected here and mixed with fine bagacillio particles for further processing.	Acid attack
Rotary vacuum filter	Filter screen, condenser, and pipelines.	Mild steel	To filter out juice and separate the impurities.	Acid attack and bace and wear corrosion
Evaporator	Calendria tube plates, tubes, shell, and save all	Mild and SS	To evaporate water from the juice.	Acidic vapor attack
Syrup sulfiter	Tank, pipelines, and SO ₂ gas outlet lines	Mild steel	Used for better mixing of SO_2 gas with the syrup for the purpose of bleaching.	SO ₂ gas attack
Pan supply tank	Rectangular tank	Mild steel	To store the syrup and then supply the same to the vacuum pan.	General corrosion
Vacuum pan	Body, calendria, vapor lines	Mild and SS	Sulfited syrup and other high-purity intermediate products are boiled to produce a mixture of sugar crystals (massecuite) and mother molasses.	SO ₂ gas attack, pitting general corrosion
Crystallizer	Air-cooled, water-cooled, vacuum, and vertical accessories	Mild steel	To reconcentrate the syrup and cool successively so as to obtain sugar crystals.	Internal corrosion in pipelines
Centrifugal	Pug mill, mixer, melt tank, and pipelines	Mild steel	The sugar crystals are separated from mother liquor.	Wear and general co
Injection and spray system	Pipelines	Mild steel	Used for recirculation of hot and cold water within the process.	Internal corrosion for low-pH water
Sugar drying and bagging unit	Multitray hopper graders	Mild steel	Sugar from centrifuge is dried in the multitray hopper, graded according to grain size in the graders, bagged, weighed, and stored in the godown.	General corrosion

placing corroded and damaged parts in most industry sectors. In this study, the authors analyzed the cost of corrosion by considering a service life of 18 years according to the guidelines of the U.S. Treasury Department (Washington, D.C.).⁵

India requires the sugar industry to pay 35% of its net income as tax. This factor also is taken into account while computing the cost of corrosion;

a 12% interest rate considers a 10% bank rate plus 4% net profit minus 2% inflation rate.

Table 2 presents data generated on 25 major components in a typical sugar factory. It indicates annual expenditure, present worth factor, tax credit, present value, annual cost factor, and equivalent annual cost—in that order. The total annual cost of corrosion per sugar factory comes to

~827,600. This figure is based facility's expenditure (cash for corrosion-related activities, spainting, maintenance, and ment. Figure 1 shows the distribution.

Directly extrapolating \$27,0 each of the country's 506 factoried at \$14 million annual accorrosion for the Indian sugartry. This is only the direct cost.

Materials Selection & Design

TABLE 3

COMPARISON OF MILD STEEL AND 409M SS

Types of	Cash Flow		Present Worth	Present	Annual Cost	Equi
Steel	Initial	Annual	Factor		Factor	C
Mild steel	-	14,127	7.25	1,02,420	0.14	1
409M SS	70,380	_	_	70,380	0.14	- 6
					Net saving per year	4



Corroded piping in a sugar factory crystallizer.

analysis, taking into account the indirect costs, probably would show a higher figure. Nearly \$2.2 million could be saved each year by using 409M SS instead of ordinary mild steel.

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References

R. Bhaskaran, N.S. Rengaswamy, N.
Palaniswamy, "Impact of Metallic Corrosion—Part II

Economic Impact," Corrosion Update, No. 60 (2001)

- G.H. Koch, M.P.H. Brongers, N.G. Thompson,
 Y.P. Virmani, J.H. Payer, "Corrosion Cost & Preventive Strategies in the United States," FHWA Report no.
 RD-01-156 (2001).
- "Depreciation—Guidelines and Rules" (Rev. Proc. 62-21), U.S. Treasury Dept. IRS Publication no. 156 (7-62), July 1962.
- S.P. Jain, "Corrosion Problem in Sugar Industries," Seventh National Congress on Corrosion Control, held Sent. 1997 (Hyderabad).

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