

High-frequency EPR investigation of X-ray storage SrBPO₅:Eu phosphor

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Abstract

High-frequency EPR (HFEP) spectra of the SrBPO₅:Eu with stillwellite structure, which was prepared using EuCl₃·6H₂O as a dopant, have been examined in order to clarify the Eu sites in it. They showed an intense doublet at 3.3 T corresponding to $g = 2.0$ as is expected for the ⁸S state ions. Also, weak satellites were observed on both sides of the central doublet signal symmetrically. This means that microwave energy at 90 GHz is sufficiently larger than the zero-field splitting. Taking into account the crystal structure of stillwellite and the earlier PL investigation, a single Eu²⁺ site was assumed for the HFEP simulation. As a result, it was found that HFEP spectrum simulated using $g = 1.98161$, $|D| = 0.130$ T and $\lambda = |E/D| = 0.06$ reasonably agrees well with the experimental one, supporting that the Eu²⁺ occupies the Sr site. The large $|D|$ value small λ value obtained implies that the crystal field surrounding Eu²⁺ has a strong axial field with weak rhombic distortion.

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1. Introduction

X-ray storage phosphors are practically useful for digital radiography in the medical diagnosis field. Recently, two of the authors have examined Eu²⁺-doped phosphor SrBPO₅:Eu with an emission at 390 nm by means of X-ray crystallography, X-band EPR spectroscopy and photoluminescence spectroscopy [1]. According to them, at low

loadings of Eu²⁺ the X-band EPR spectra showed 12 lines centred around $g = 2.0$, which suggests that Eu²⁺ occupies a high-symmetry site. At higher loadings, however, very complicated EPR resonance, which is not simply assigned to Eu²⁺ in a single site, was observed, implying that some Eu²⁺ occupy low-symmetry site(s).

Previously we have succeeded in identifying low-symmetry Eu²⁺ which is doped in alkaline earth aluminate phosphors by means of high-frequency EPR (HFEP) spectroscopy [2–5], and confirmed that the HFEP has advantage over X-band EPR

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for the investigation of metal ions with a large zero-field splitting. In this paper, therefore, we report the results of the HFEPR investigation of the SrBPO₅:Eu phosphor.

2. Experimental

Eu²⁺-doped samples were prepared by conventional solid-state reaction with SrCO₃, H₃BO₃, (NH₄)HPO₄ and EuCl₃·6H₂O (or Eu₂O₃) under mild reducing atmosphere. Eu²⁺ concentration was adjusted to 1 mol%, because in the previous study the optimum europium(II) concentration for photoluminescence and photo-stimulated luminescence is 1 mol% [1]. The preparation procedure was described elsewhere [1]. HFEPR measurements at 90 GHz were carried out at room temperature using a novel quasi-optical induction mode spectrometer that was designed and built at the University of St. Andrews, UK.

3. Results and discussion

Fig. 1 shows the X-band EPR spectrum of the SrBPO₅:Eu prepared with a EuCl₃ of 1.0 mol%. The same spectral pattern was obtained when

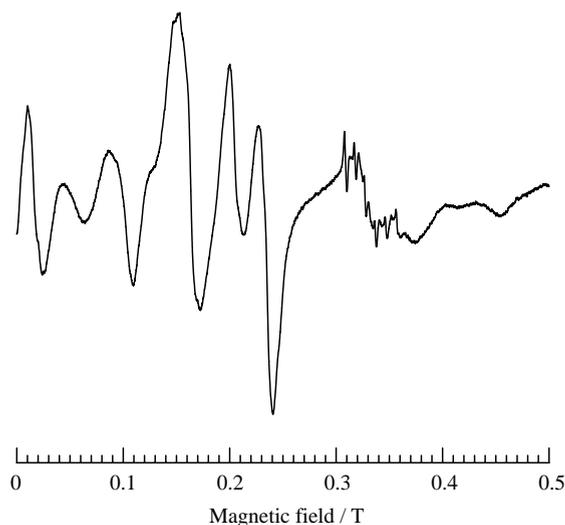


Fig. 1. X-band EPR spectrum of the SrBPO₅:Eu with stillwellite structure recorded at room temperature.

Eu₂O₃ was used as a dopant, implying that the chloride ion rarely affects the environment surrounding Eu²⁺. Also, it was confirmed that the EPR spectrum obtained in this study is fundamentally the same as those reported by one of the authors earlier, although the resonance attributed to Eu²⁺ in a high-symmetry site is not obvious because of the intense hyperfine lines due to Mn²⁺ impurity.

It is seen that several broad signals appear over the wide range of magnetic field. Amongst them the signals appearing in the magnetic field in the range of 0–0.25 T are stronger than those observed at about 0.33 T although the *g*-value of ⁸S state ions without orbital angular momentum should be close to 2.0. Such X-band EPR spectrum is a typical pattern for the Eu²⁺ with a large zero-field splitting, which results in the mixing of the wave functions. In other words, Eu²⁺ occupies low-symmetry site(s) [2–5].

Fig. 2(a) shows the 90 GHz HFEPR spectrum of the SrBPO₅:Eu prepared using EuCl₃·6H₂O as a

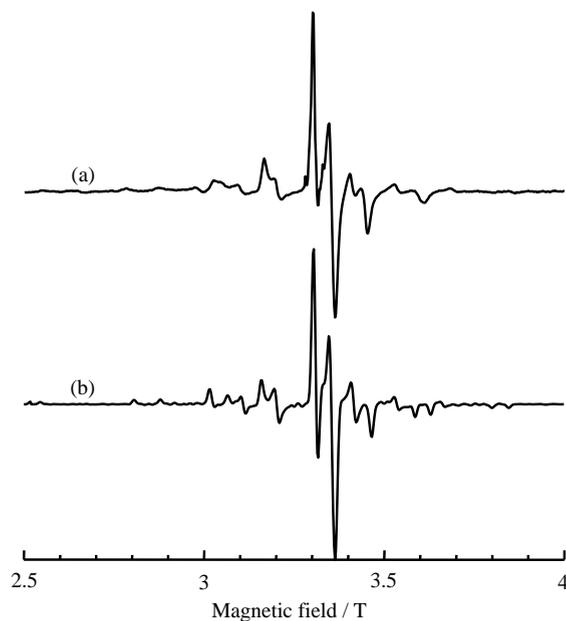


Fig. 2. Room-temperature 90 GHz EPR spectra of the SrBPO₅:Eu with stillwellite structure: (a) experimental spectrum and (b) simulated spectrum with $g = 1.9816$, $|D| = 0.130$ T and $\lambda = |E/D| = 0.06$. Lorentzian line shape with a peak-to-peak width of $\Delta H = 0.0075$ T was assumed.

dopant. It is immediately noticed that the obtained HFEP spectrum is different from that measured at X-band. An intense doublet signal appears at 3.3 T corresponding to $g = 2$ as is expected for the 8S state ions. Also, weak satellites are observed on both sides of the doublet signal symmetrically. When the HFEP spectrum is measured under the condition satisfying Zeeman energy \gg zero-field splitting energy, it consists of seven fine structure lines for $\Delta M_S = \pm 1$. This means that microwave energy at 90 GHz is large enough compared with the zero-field splitting.

Based on the crystal structure of stillwellite [6] and the results of the photoluminescence investigation reported earlier [1] a single Eu^{2+} site is assumed for the HFEP simulation. EPR parameters were roughly estimated as follows: g -value was calculated using $g = hv/\beta H$ with the magnetic field where the central intense signal crosses zero in intensity. From the satellites appearing at 3.19 and 3.45 T next to the intense doublet the $|D|$ value is 0.15 T. For the $|E/D|$ ($= \lambda$) an initial value of 0.1 was used. This is because three absorption-like satellites at $\lambda = \frac{1}{3}$ should be expected on both sides of the central signal, while three absorption-like and three derivative-like satellites at $\lambda = 0$. Observed HFEP spectrum implies that the λ value is in between.

Simulated HFEP spectrum, which was carried out using the equation derived using the second perturbation theory [7,8], is shown in Fig. 2(b). The parameters used were $g = 1.9816$, $|D| = 0.130$ T and $\lambda = |E/D| = 0.06$. It is obvious that the simulated HFEP spectrum reasonably agrees well with the experimental one, supporting that the Eu^{2+} occupies the strontium site since there is only a single site in the unit cell of stillwellite. The obtained $|D|$ value is very large, while the λ value is small, implying that the crystal field surrounding Eu^{2+} has a strong axial field with weak rhombic distortion.

From the photoluminescence studies [1], we observe that the stillwellite-type SrBPO_5 system, for low levels of Eu^{2+} doping ($\sim 0.01\%$) shows a weak emission band around 350 nm while for moderate level of Eu^{2+} doping (1–5%) an intense narrow band around 390 nm is observed. These two emission bands might suggest two kinds of

cationic site(s) occupied by Eu^{2+} in this system. The absence of 350 nm emission band for higher levels of Eu^{2+} doping may be explained in terms of intersite energy transfer in favour of 390 nm emission band having profound dependence on Eu^{2+} concentration. For all practical purposes, the intense 390 nm emission band due to Eu^{2+} centre in SrBPO_5 can be considered to represent Eu^{2+} emission in SrBPO_5 . The band emission from Eu^{2+} centre(s) can be assigned to $4f^65d \rightarrow ^8S_{7/2}$ (an allowed f–d type) transition. The occurrence of band emission due to parity allowed ‘f–d’ transition that is possible irrespective of whether the cationic site(s) occupied by Eu^{2+} has an inversion centre or not and hence symmetry independent.

The narrow band emission observed from Eu^{2+} centre(s) clearly point out that the lowest excited state is $4f^65d$ state. The 5d orbital of Eu^{2+} can undergo considerable crystal field splitting determined by the symmetry and the chemical coordination of the cationic site(s) it occupies.

In conclusion, it is found that for the $\text{SrBPO}_5:\text{Eu}$ the Eu^{2+} occupies the strontium site which has strong axial crystal field with weak rhombic distortion. This explains the PL behaviour as follows: A simple comparison of the excitation spectrum of $\text{SrBPO}_5:\text{Eu}^{2+}$ in relation to cubic $\text{KCl}:\text{Eu}^{2+}$ readily reveals that the excitation spectrum due to the splitting of the 5d orbital in SrBPO_5 is more complex and broader by approximately 20%. This suggests that the axial field parameter $|D| = B_2^0$ is much larger in SrBPO_5 . Further, the complex excitation spectrum of $\text{SrBPO}_5:\text{Eu}^{2+}$ implies that this cationic site can have weak rhombic field too (low value for E). These results are possible for Eu^{2+} occupying a 9 or 10 coordinated sites due to the fact that Sr^{2+} has no symmetry elements [9,10]. These results are in good agreement with the conclusions deduced from the HFEP studies on $\text{SrBPO}_5:\text{Eu}^{2+}$.

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References

- [1] A. Karthikeyani, R. Jagannathan, *J. Lumin.* 86 (2000) 79.
- [2] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *J. Chem. Soc. Faraday Trans.* 94 (1998) 3009.
- [3] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *Phys. Chem. Chem. Phys.* 1 (1999) 4011.
- [4] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, C.C. Rowlands, V. Beltran-Lopez, G.M. Smith, P.C. Riedi, *J. Mater. Chem.* 10 (2000) 2566.
- [5] T. Nakamura, K. Kaiya, N. Takahashi, T. Matsuzawa, M. Ohta, C.C. Rowlands, G.M. Smith, P.C. Riedi, *Phys. Chem. Chem. Phys.* 3 (2001) 1721.
- [6] A.A. Voronkov, Yu.A. Pyatenko, *Sov. Phys. Cryst.* 12 (1967) 214.
- [7] J.R. Pilbrow, *Transition Ion Electron Paramagnetic Resonance*, Clarendon Press, Oxford, 1990, p. 614.
- [8] T. Takeyama, T. Nakamura, T. Takahashi, V. Beltran-Lopez, C.C. Rowlands, in: A. Kawamori, J. Yamauchi, H. Ohta (Eds.), *EPR in the 21st Century*, Elsevier Science B.V., Netherlands, 2001, 63pp.
- [9] Y. Ono, K. Takayanama, T. Kajitani, *J. Phys. Soc. Jpn.* 65 (1996) 3224.
- [10] A.A. Voronkov, Yu.A. Pyatenko, *Kristallografiya* 12 (1967) 135.